# Empirical formulae for calculating Y -ray detectors effective solid angle ratio 

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## ABSTRACT

Determination of the detector efficiency using volumetric cylindrical sources is very important in various scientific and industrial fields, especially in the field of quantitative analysis. To calculate the absolute activity of any sample, the full-energy peak efficiency (FEPE) of the detector is needed. By applying the efficiency transfer method, the FEPE of the detector would be determined easily without using the standard sources. This approach depends on two main factors. The first one, is the reference efficiency of the reference source, which is determined experimentally, and the second one, is the calculation of the effective solid angle ratio between the sample and the reference source geometries. This work introduces an empirical formula for calculating the second factor for using two different sizes of $\mathrm{NaI}(\mathrm{TI})$ detectors. The validity of this empirical formula was successfully demonstrated by comparing the calculating values with the experimental values.

## INTRODUCTION

The scintillation counters are used to measure the radiation in different applications such as, radiation survey meters, medical imaging, nuclear plant safety, measuring radon levels, oil well logging and monitoring for radioactive contamination. In the gamma-ray spectroscopy, one usually needs to know the full-energy peak efficiency for any specific source-to-detector configuration of concern. Traditionally, measurements are performed in gamma-ray spectrometry by the relative method, according to which the measured sample is first prepared, that should match the used standard source in all the important characteristics, such as its size, chemical composition and density [1]. This method is tedious and time consuming process. In order to overcome the problems of the experimental method, several non-experimental methods [2-6] have been proposed and applied, depending on the photon energy, source-to-detector geometry and volume. One of the most common approaches is called the efficiency transfer method. In this technique, the detector efficiency of using various source dimensions is derived from the known efficiency for the reference source-to-detector geometry. The efficiency transfer method is particularly useful due to, its insensitivity to the
inaccuracy of the input data, such as the uncertainty of the detector characterization [7,8].

Badawi, et al. [9-11] were introduced an approach to calculate the full-energy peak efficiency for $\mathrm{NaI}(\mathrm{Tl})$ and HPGe detectors, with respect to different volumetric sources. This approach stated that, the detector efficiency using a certain cylindrical radioactive source, $\varepsilon(\mathrm{E}, \mathrm{Cyl})$, equal the reference efficiency of using reference radioactive point source, $\varepsilon\left(E, P_{i}\right)$, with the same detector multiplied by the effective solid angle ratio, $R$, between the two geometries and expressed by the following equation

$$
\begin{equation*}
\varepsilon(\mathrm{E}, \mathrm{Cyl})=\varepsilon\left(\mathrm{E}, \mathrm{P}_{\mathrm{i}}\right) \cdot \mathrm{R} \tag{1}
\end{equation*}
$$

Calculations of the effective solid angle are based on the direct mathematical method which reported by Selim and Abbas [12-16] and used successfully before to calibrate different detectors with different sources. The present work will introduce empirical equations to calculate the effective solid angle ratios of two $\mathrm{NaI}(\mathrm{Tl})$ detectors with different geometries. The effective solid angle ratio can be used as a conversion factor from using the radioactive point source case to the case in which the cylindrical radioactive sources were used. Consequently, the corresponding full-energy peak efficiency can be calculated simply.

## EXPERIMENTAL SETUP

The full-energy peak efficiency (FEPE) values were determined for two NaI (Tl) detectors with resolutions $8.5 \%$ and $7.5 \%$ at the 662 keV peaks of ${ }^{137} \mathrm{Cs}$ labeled as D1 and D2 respectively. The manufacturer parameters and the setup values are shown in table 1. The experimental measurements were carried out by using point and cylindrical radioactive sources.

The radioactive standard point sources $\left({ }^{241} \mathrm{Am},{ }^{133} \mathrm{Ba},{ }^{152} \mathrm{Eu},{ }^{137} \mathrm{Cs}\right.$, and $\left.{ }^{60} \mathrm{Co}\right)$ are used for the calibration of gamma spectrometers. The radioactive substance is a very thin, compact grained layer applied to a circular area about 5 mm in diameter, in the middle of the source between two polyethylene foils and each having a mass per unit area of ( $21.3 \pm 1.8$ ) mg. $\mathrm{cm}^{-2}$. By heating under pressure, the two foils are welded together over the whole area so that they are leak-proofed. To facilitate handling, the foil 26 mm in diameter is mounted in a circular aluminum ring (outer diameter: 30 mm , height: 3 mm ) from which it can easily be removed if and when required. These point sources were purchased from the Physikalisch-Technische Bundesanstalt (PTB) in Braunschweig and Berlin, which is the national institute for science and technology and the highest technical authority of the Federal Republic of Germany in the field of metrology and certain sectors of safety engineering. The sources activities and their uncertainties, half-lives, photon energies, and photon emission probabilities per decay for all of PTB sources are listed in table 2.

The homemade Plexiglas holder was used to measure these standard point sources, each at seven different axial distances starting from 20 cm up to 50 cm from the surface of the detector (with a 5 cm as a step). The measurements started from a source-todetector distance equals 20 cm to minimize the effect of the coincidence summing effect. Spectra were recorded as, P4D1, where P refers to the source type (point) measured at the detector (D1) at position number (4), which equal 20 cm .

The cylindrical radioactive sources were in polypropylene plastic vials form with radius greater than the radius of the detectors, and volumes of $200 \mathrm{ml}, 300 \mathrm{ml}$ and 400 ml filled with an aqueous solution containing ${ }^{152} \mathrm{Eu}$ radionuclide, which used for the calibration process. The ${ }^{152} \mathrm{Eu}$ source emits $\gamma$-ray in the energy range from 121.78 keV up to 1408.01 keV . Table 3 shows the source dimensions. In order to minimize the dead

Table 1: The manufacturer parameters and the setup values.

| Items | Detector (D1) | Detector (D2) |
| :---: | :---: | :---: |
| Manufacturer | Canberra | Canberra |
| Serial Number | 09L 654 | 09L 652 |
| Detector Model | 802 | 802 |
| Type | Cylindrical | Cylindrical |
| Mounting | Vertical | Vertical |
| Resolution (FWHM) at 661 keV | 7.5\% | 8.5\% |
| Cathode to Anode voltage | +900 V dc | $+800 \mathrm{Vdc}$ |
| Dynode to Dynode | +80 V dc | +80 V dc |
| Cathode to Dynode | +150 V dc | +150 V dc |
| Tube Base | Model 2007 | Model 2007 |
| Shaping Mode | Gaussian | Gaussian |
| Detector Type | $\mathrm{NaI}(\mathrm{TI})$ | $\mathrm{NaI}(\mathrm{TI})$ |
| Crystal Diameter (mm) | 50.8 | 76.2 |
| Crystal Length (mm) | 50.8 | 76.2 |
| Top cover Thickness (mm) | Al (0.5) | Al (0.5) |
| Side cover Thickness (mm) | Al (0.5) | Al (0.5) |
| Reflector - Oxide (mm) | 2.5 | 2.5 |
| Weight ( Kg ) | 0.77 | 1.8 |
| Outer Diameter (mm) | 57.2 | 80.9 |
| Outer Length (mm) | 53.9 | 79.4 |
| Crystal Volume in ( $\mathrm{cm}^{3}$ ) | 103.004 | 347.639 |

Table 2: Point sources activities and their uncertainties, half lives, photon energies and photon emission probabilities per decay for the all radionuclides used in this work.

| PTB Nuclide | Energy (keV) | Emission Probability \% | Half Life <br> (Days) | Activity (kBq) <br> At 1.June 2009 00:00 Hr | Uncertainty (KBq) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| ${ }^{241} \mathrm{Am}$ | 59.52 | 35.9 | 157861.05 | 259.0 | $\pm 2.6$ |
| ${ }^{133} \mathrm{Ba}$ | 80.99 | 34.1 | 3847.91 | 275.3 | $\pm 2.8$ |
| ${ }^{152} \mathrm{Eu}$ | 121.78 | 28.4 | 4943.29 | 290.0 | $\pm 4.0$ |
|  | 244.69 | 7.49 |  |  |  |
|  | 344.28 | 26.6 |  |  |  |
|  | 778.91 | 12.96 |  |  |  |
|  | 964.13 | 14.0 |  |  |  |
|  | 1408.01 | 20.87 |  |  |  |
| ${ }^{137} \mathrm{Cs}$ | 661.66 | 85.21 | 11004.98 | 385.0 | $\pm 4.0$ |
| ${ }^{60} \mathrm{Co}$ | 1173.23 | 99.99 | 1925.31 | 212.1 | $\pm 1.5$ |
|  | 1332.51 | 99.98 |  |  |  |

Table 3: Parameters of the radioactive cylindrical volumetric sources.

| Items | Source Volume (ml) |  |  |
| :---: | :---: | :---: | :---: |
|  | V1=200 | V2=300 | V3=400 |
| Inner diameter (mm) |  | 111.50 |  |
| Height (mm) | 21.45 | 31.59 | 41.83 |
| Wall and Bottom thickness (mm) |  | 2.03 |  |
| Activity (Bq) |  | $5048 \pm 49.98$ |  |
| At 1.Jan 2010 00:00 Hr |  |  |  |

time, the activity of the sources is prepared to be a few kilo Becquerel $(5048 \pm 49.98$ $B q)$.

The radioactive volumetric cylindrical sources were measured on a 0.36 cm thickness Plexiglas cover and placed directly on the detector end-cap. These measurements were done using two cylindrical detectors with numbers (D1 \& D2). Figure 1 shows a diagram


Figure 1: A diagram of a cylindrical detector with radioactive cylindrical source.
of a cylindrical detector with cylindrical source. Spectra were recorded as V1D2, where V1 is the volume (V1) measured at the detector (D2). The angular correlation effects can be neglected for the low source-to-detector distance $[17,18]$.

All the measurements are carried out to obtain statistically significant main peaks in the spectra that are recorded and processed by winTMCA32 software made by ICx Technologies. Measured spectrum, which saved as spectrum ORTEC files can be opened by the Genie 2000 data acquisition and analysis software made by Canberra. The acquisition time is high enough to get at least the number of counts 20,000, which make the statistical uncertainties less than $0.1 \%$. The spectra are analyzed with the program using its automatic peak search and peak area calculations, along with changes in the peak fit using the interactive peak fit interface when necessary to reduce the residuals and error in the peak area values. The peak areas, the live time, the run time and the start time for each spectrum were entered in the spreadsheets that are used to perform the calculations necessary to generate the efficiency curves.

## RESULTS AND DISCUSSIONS

The efficiency transfer theoretical method (ETTM) has been used to convert the (FEPE) curve for using radioactive point source at positions start from P4 up to P10 to the (FEPE) for using radioactive cylindrical sources, which represented in V1, V2, and V3. These calculations extended for two cylindrical $\mathrm{NaI}(\mathrm{Tl})$ detectors (D1 \& D2). By using equation (1) and the experimental efficiency values for using point and cylindrical radioactive sources, that published before in 2012 [19], the one can calculate the effective solid angle ratio, R , values for both detectors experimentally as tabulated in table 4.

The analytical expressions presented in [19] were used to calculate the effective solid angle ratio as presented in table 5, these values were tested before to obtain the detector FEPE and it was accepted by comparison with the experimental values. The percentage deviations between the effective solid angle ratio values obtained by the two methods are shown in figure 2. A remarkable agreement between them was achieved with discrepancies less than $10 \%$.

By plotting a three dimensional relation between the Log values of the point source position, $\mathrm{P}(\mathrm{cm})$, the effective solid angle ratio, R , and the photon energy, $\mathrm{E}(\mathrm{keV})$ for the two detectors (D1 \& D2) was done as shown in figure 3. The plotted data for each

Table 4: The values of the effective solid angle ratio, R , for both detectors, which were obtained experimentally.
Effective solid angle ratio between different volumes and different positions with respect to (D1 and D2).(From Experimental Data)
Detector (D1) Effective solid angle ratio
Detector (D2) Effective solid angle ratio

| Nuclide | Energy | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 4}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{V} 1}}{\Omega_{\mathrm{P} 7}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{V} 1}}{\Omega_{\mathrm{P} 9}}$ | $\frac{\Omega_{\mathrm{V} 1}}{\Omega_{\mathrm{P} 10}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 4}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{V} 1}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 9}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 10}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Eu-152 | 121.78 | 15.722 | 23.735 | 32.957 | 44.330 | 58.034 | 72.859 | 89.042 | 11.984 | 17.753 | 25.185 | 33.687 | 44.037 | 54.425 | 66.609 |
| Eu-152 | 244.69 | 15.795 | 23.603 | 33.259 | 44.087 | 57.531 | 72.710 | 87.406 | 12.380 | 18.381 | 25.950 | 34.453 | 45.196 | 55.989 | 68.073 |
| Eu-152 | 344.28 | 15.850 | 23.578 | 33.370 | 44.110 | 57.503 | 73.168 | 87.516 | 12.572 | 18.519 | 26.253 | 34.492 | 45.250 | 56.562 | 68.579 |
| Eu-152 | 778.9 | 16.165 | 24.068 | 34.135 | 45.149 | 58.498 | 74.488 | 89.140 | 12.935 | 19.337 | 26.901 | 35.421 | 45.992 | 57.317 | 70.413 |
| Eu-152 | 964.13 | 16.298 | 24.132 | 34.285 | 45.482 | 58.722 | 74.473 | 89.618 | 13.070 | 19.444 | 27.349 | 35.753 | 46.408 | 57.349 | 70.154 |
| Eu-152 | 1408.01 | 16.465 | 24.458 | 34.752 | 45.831 | 59.643 | 74.802 | 89.715 | 13.182 | 19.596 | 27.499 | 36.195 | 46.989 | 58.306 | 71.399 |
| Nuclide | Energy | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 8}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 10}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{V} 2}}{\Omega_{\mathrm{P} 9}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 10}}$ |
| Eu-152 | 121.78 | 13.216 | 19.951 | 27.703 | 37.263 | 48.782 | 61.244 | 74.847 | 10.313 | 15.278 | 21.674 | 28.990 | 37.897 | 46.837 | 57.322 |
| Eu-152 | 244.69 | 13.222 | 19.758 | 27.841 | 36.905 | 48.159 | 60.864 | 73.166 | 10.572 | 15.697 | 22.161 | 29.423 | 38.598 | 47.815 | 58.134 |
| Eu-152 | 344.28 | 13.372 | 19.892 | 28.153 | 37.214 | 48.514 | 61.730 | 73.835 | 10.679 | 15.731 | 22.300 | 29.298 | 38.437 | 48.045 | 58.253 |
| Eu-152 | 778.9 | 13.755 | 20.480 | 29.046 | 38.418 | 49.777 | 63.383 | 75.851 | 11.215 | 16.767 | 23.326 | 30.713 | 39.878 | 49.699 | 61.054 |
| Eu-152 | 964.13 | 13.911 | 20.597 | 29.263 | 38.820 | 50.121 | 63.565 | 76.491 | 11.303 | 16.814 | 23.650 | 30.918 | 40.132 | 49.593 | 60.666 |
| Eu-152 | 1408.01 | 14.095 | 20.938 | 29.749 | 39.234 | 51.058 | 64.035 | 76.802 | 11.435 | 16.999 | 23.855 | 31.398 | 40.761 | 50.579 | 61.937 |
| Nuclide | Energy | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 10}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\mathrm{P} 9}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 10}}$ |
| Eu-152 | 121.78 | 11.152 | 16.835 | 23.377 | 31.444 | 41.164 | 51.679 | 63.158 | 8.760 | 12.977 | 18.410 | 24.624 | 32.190 | 39.783 | 48.690 |
| Eu-152 | 244.69 | 11.372 | 16.994 | 23.947 | 31.743 | 41.423 | 52.352 | 62.933 | 9.009 | 13.376 | 18.884 | 25.072 | 32.889 | 40.743 | 49.536 |
| Eu-152 | 344.28 | 11.483 | 17.082 | 24.176 | 31.957 | 41.661 | 53.010 | 63.405 | 9.167 | 13.504 | 19.143 | 25.150 | 32.995 | 41.243 | 50.005 |
| Eu-152 | 778.9 | 11.901 | 17.720 | 25.132 | 33.242 | 43.070 | 54.842 | 65.630 | 9.582 | 14.325 | 19.929 | 26.241 | 34.072 | 42.462 | 52.164 |
| Eu-152 | 964.13 | 12.024 | 17.803 | 25.294 | 33.553 | 43.321 | 54.942 | 66.115 | 9.878 | 14.695 | 20.669 | 27.021 | 35.073 | 43.342 | 53.020 |
| Eu-152 | 1408.01 | 12.287 | 18.251 | 25.933 | 34.201 | 44.508 | 55.820 | 66.949 | 9.997 | 14.861 | 20.855 | 27.450 | 35.635 | 44.219 | 54.148 |

Table 5: The values of the effective solid angle ratio, R, for both detectors, which are obtained analytically [19].
Effective solid angle ratio between different volumes and different positions with respect to (D1 and D2).
Detector (D1) Effective solid angle ratio
Detector (D2) Effective solid angle ratio

| Nuclide | Energy | $\frac{\Omega_{\mathrm{v}_{1}}}{\Omega_{\mathrm{P} 4}}$ | $\frac{\Omega_{\mathrm{v}_{1}}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v}_{1}}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{p} 8}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 10}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v}_{1}}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v}_{1}}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 10}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Am-241 | 59.53 | 14.637 | 21.861 | 31.549 | 40.638 | 53.398 | 66.338 | 83.377 | 11.666 | 17.293 | 24.512 | 32.344 | 42.805 | 51.785 | 62.635 |
| Ba-133 | 80.99 | 14.903 | 22.260 | 31.895 | 41.453 | 54.488 | 68.058 | 84.158 | 11.703 | 17.359 | 24.550 | 32.436 | 42.850 | 52.310 | 63.409 |
| Eu-152 | 121.78 | 15.209 | 22.720 | 32.416 | 42.347 | 55.684 | 69.808 | 85.445 | 11.867 | 17.612 | 24.869 | 32.889 | 43.394 | 53.301 | 64.747 |
| Eu-152 | 244.69 | 15.779 | 23.573 | 33.496 | 43.962 | 57.849 | 72.828 | 88.196 | 12.293 | 18.254 | 25.728 | 34.071 | 44.887 | 55.523 | 67.663 |
| Eu-152 | 344.28 | 16.080 | 24.024 | 34.078 | 44.815 | 58.992 | 74.403 | 89.687 | 12.528 | 18.608 | 26.206 | 34.725 | 45.718 | 56.725 | 69.230 |
| Cs-137 | 661.66 | 16.633 | 24.852 | 35.145 | 46.375 | 61.084 | 77.302 | 92.414 | 12.965 | 19.266 | 27.093 | 35.941 | 47.263 | 58.969 | 72.169 |
| Eu-152 | 778.9 | 16.766 | 25.051 | 35.401 | 46.751 | 61.588 | 77.999 | 93.070 | 13.070 | 19.424 | 27.307 | 36.233 | 47.635 | 59.509 | 72.876 |
| Eu-152 | 964.13 | 16.933 | 25.302 | 35.724 | 47.224 | 62.223 | 78.880 | 93.897 | 13.203 | 19.625 | 27.577 | 36.603 | 48.104 | 60.193 | 73.773 |
| Co-60 | 1173.23 | 17.081 | 25.523 | 36.010 | 47.642 | 62.783 | 79.658 | 94.628 | 13.322 | 19.803 | 27.818 | 36.933 | 48.523 | 60.802 | 74.571 |
| Co-60 | 1332.5 | 17.175 | 25.663 | 36.191 | 47.907 | 63.138 | 80.149 | 95.090 | 13.397 | 19.916 | 27.969 | 37.141 | 48.788 | 61.186 | 75.074 |
| Eu-152 | 1408.01 | 17.216 | 25.724 | 36.270 | 48.022 | 63.292 | 80.363 | 95.293 | 13.429 | 19.964 | 28.034 | 37.230 | 48.900 | 61.350 | 75.290 |
| Nuclide | Energy | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 4}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 10}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 4}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 9}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 10}}$ |
| Am-241 | 59.53 | 12.081 | 18.044 | 26.041 | 33.544 | 44.076 | 54.757 | 68.821 | 9.926 | 14.714 | 20.855 | 27.519 | 36.419 | 44.060 | 53.292 |
| Ba-133 | 80.99 | 12.365 | 18.470 | 26.464 | 34.394 | 45.210 | 56.469 | 69.827 | 10.011 | 14.851 | 21.002 | 27.749 | 36.657 | 44.751 | 54.246 |
| Eu-152 | 121.78 | 12.691 | 18.959 | 27.050 | 35.336 | 46.466 | 58.251 | 71.300 | 10.214 | 15.159 | 21.405 | 28.308 | 37.350 | 45.877 | 55.728 |
| Eu-152 | 244.69 | 13.301 | 19.872 | 28.238 | 37.060 | 48.767 | 61.394 | 74.350 | 10.698 | 15.886 | 22.391 | 29.651 | 39.064 | 48.320 | 58.886 |
| Eu-152 | 344.28 | 13.628 | 20.361 | 28.882 | 37.981 | 49.996 | 63.057 | 76.010 | 10.966 | 16.288 | 22.938 | 30.395 | 40.018 | 49.652 | 60.598 |


| Cs-137 | 661.66 | 14.232 | 21.265 | 30.072 | 39.682 | 52.268 | 66.145 | 79.076 | 11.467 | 17.039 | 23.962 | 31.787 | 41.800 | 52.153 | 63.828 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Eu-152 | 778.9 | 14.378 | 21.484 | 30.360 | 40.094 | 52.818 | 66.893 | 79.817 | 11.588 | 17.221 | 24.210 | 32.124 | 42.232 | 52.760 | 64.611 |
| Eu-152 | 964.13 | 14.563 | 21.761 | 30.725 | 40.615 | 53.51 | 67.841 | 80.756 | 11.742 | 17.452 | 24.524 | 32.551 | 42.779 | 53.529 | 65.606 |
| Co-60 | 1173.23 | 14.727 | 22.006 | 31.048 | 41.07 | 54.131 | 68.680 | 81.587 | 11.879 | 17.658 | 24.804 | 32.932 | 43.267 | 54.216 | 66.493 |
| Co-60 | 1332.5 | 14.831 | 22.161 | 31.252 | 41.369 | 54.522 | 69.212 | 82.114 | 11.966 | 17.788 | 24.982 | 33.173 | 43.576 | 54.649 | 67.054 |
| Eu-152 | 1408.01 | 14.877 | 22.229 | 31.342 | 41.497 | 54.693 | 69.445 | 82.346 | 12.003 | 17.844 | 25.058 | 33.277 | 43.709 | 54.837 | 67.297 |
| Nuclide | Energy | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\mathrm{P} 4}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 7}}$ | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 10}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 4}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\text {P6 }}}$ | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\text {P7 }}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 8}}$ | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\mathrm{P} 9}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 10}}$ |
| Am-241 | 59.53 | 10.175 | 15.197 | 21.932 | 28.251 | 37.12 | 46.117 | 57.962 | 8.322 | 12.336 | 17.486 | 23.073 | 30.536 | 36.942 | 44.682 |
| Ba-133 | 80.99 | 10.459 | 15.622 | 22.38 | 29.09 | 38.24 | 47.763 | 59.062 | 8.425 | 12.498 | 17.675 | 23.353 | 30.850 | 37.661 | 45.652 |
| Eu-152 | 121.78 | 10.785 | 16.112 | 22.98 | 30.02 | 39.48 | 49.503 | 60.592 | 8.632 | 12.811 | 18.089 | 23.923 | 31.564 | 38.770 | 47.095 |
| Eu-152 | 244.69 | 11.400 | 17.032 | 24.202 | 31.763 | 41.79 | 52.620 | 63.724 | 9.110 | 13.527 | 19.066 | 25.249 | 33.264 | 41.145 | 50.142 |
| Eu-152 | 344.28 | 11.733 | 17.529 | 24.865 | 32.699 | 43.04 | 54.288 | 65.440 | 9.375 | 13.925 | 19.611 | 25.986 | 34.212 | 42.449 | 51.807 |
| Cs-137 | 661.66 | 12.354 | 18.458 | 26.103 | 34.44 | 45.369 | 57.414 | 68.638 | 9.874 | 14.673 | 20.635 | 27.373 | 35.996 | 44.911 | 54.964 |
| Eu-152 | 778.9 | 12.505 | 18.684 | 26.404 | 34.870 | 45.936 | 58.177 | 69.417 | 9.996 | 14.855 | 20.884 | 27.711 | 36.430 | 45.512 | 55.734 |
| Eu-152 | 964.13 | 12.697 | 18.972 | 26.787 | 35.410 | 46.656 | 59.146 | 70.406 | 10.151 | 15.087 | 21.201 | 28.140 | 36.982 | 46.276 | 56.716 |
| Co-60 | 1173.23 | 12.868 | 19.227 | 27.127 | 35.890 | 47.296 | 60.008 | 71.285 | 10.289 | 15.294 | 21.484 | 28.524 | 37.475 | 46.958 | 57.592 |
| Co-60 | 1332.5 | 12.976 | 19.389 | 27.343 | 36.195 | 47.702 | 60.554 | 71.842 | 10.376 | 15.425 | 21.663 | 28.767 | 37.788 | 47.390 | 58.147 |
| Eu-152 | 1408.01 | 13.024 | 19.461 | 27.438 | 36.329 | 47.881 | 60.795 | 72.089 | 10.414 | 15.482 | 21.741 | 28.872 | 37.923 | 47.578 | 58.388 |



Figure 2a: The deviation between the calculated effective solid angle ratio, R, that obtained analytically and the experimental one for D1.


Figure 2b: The deviation between the calculated effective solid angle ratio, $R$, that obtained analytically and the experimental one for D2.


Figure 3: The relation between Log values of radioactive point source positions, P, solid angle ratio, R, and the photon energy, $E$, for D1 and D2.
source volume ( ml ) with the two detectors have shown semi plane shape and the empirical formulae that represent these shapes are described below to calculate the effective solid angle ratios, R, for both detectors.

The empirical formula for the detector (D1) is given by:

$$
\begin{equation*}
\log (E)-26.77 \log (R)+49.18 \log (P)-0.0176 V-30.62=0 \tag{2}
\end{equation*}
$$

while, the empirical formula for the detector (D2) is given by:

$$
\begin{equation*}
\log (\mathrm{E})-26.77 \log (\mathrm{R})+49.18 \log (\mathrm{P})-0.0166 \mathrm{~V}-33.63=0 \tag{3}
\end{equation*}
$$

By knowing the photon energy and the reference position, the effective solid angle ratio, $R$, for both detectors was calculated using equations (2) and (3). The obtained values were tabulated in table 6 . Therefore, these equations provide a simple method to calculate the full-energy peak efficiency (FEPE) of two different cylindrical $\mathrm{NaI}(\mathrm{Tl})$ scintillation detectors. These two formulae are valid through a wide energy range and different radioactive volumetric source geometries. The percentage deviations between the calculated effective solid angle ratio, that obtained experimentally and that obtained from equations (2) and (3) were shown in figure 4. A remarkable agreement between them was achieved with discrepancies less than $7 \%$.

The main advantage of this process is the simplicity of obtaining the effective solid angle ratios, R , especially in between any two measured positions, without using analytical or experimental calculations. These ratios are considered to be the efficiency conversion factor between any two different geometrical conditions, and used to save the time in absent the standard calibration sources.

## CONCLUSIONS

The present work leads to a simplified method to calculate the effective solid angle ratio empirical, which can be used to calculate the conversion factors of the detector efficiency, in the case of using point and cylindrical radioactive sources. The efficiencies can be determined at any calibration position or any energy situated in the domain of the study based on these conversion factors. These formulas are valid through a wide energy range and different source-to-detector geometries. Therefore the corresponding full-energy peak efficiency can be calculated simply, and the activity of unknown samples measured in the same conditions can be determined easily.

Table 6: The values of the effective solid angle ratio, R , for both detectors, which are obtained from empirical equations.
Effective solid angle ratio between different volumes and different positions with respect to (D1 and D2).(from the empirical formula)
Detector (D1) Effective solid angle ratio
Detector (D2) Effective solid angle ratio

| Nuclide | Energy | $\frac{\Omega_{\mathrm{v}_{1}}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v}_{1}}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v}_{1}}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 10}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v}_{1}}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v}_{1}}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{V} 1}}{\Omega_{\mathrm{P} 7}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v} 1}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{V} 1}}{\Omega_{\mathrm{P} 10}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Am-241 | 59.53 | 15.148 | 22.820 | 31.896 | 42.334 | 54.099 | 67.163 | 81.501 | 11.893 | 17.916 | 25.042 | 33.236 | 42.473 | 52.730 | 63.987 |
| Ba-133 | 80.99 | 15.323 | 23.084 | 32.265 | 42.823 | 54.725 | 67.940 | 82.444 | 12.030 | 18.123 | 25.331 | 33.621 | 42.965 | 53.340 | 64.727 |
| Eu-152 | 121.78 | 15.558 | 23.438 | 32.760 | 43.480 | 55.564 | 68.982 | 83.709 | 12.215 | 18.402 | 25.720 | 34.137 | 43.624 | 54.159 | 65.721 |
| Eu-152 | 244.69 | 15.969 | 24.057 | 33.625 | 44.628 | 57.031 | 70.804 | 85.919 | 12.537 | 18.887 | 26.399 | 35.038 | 44.776 | 55.588 | 67.456 |
| Eu-152 | 344.28 | 16.174 | 24.366 | 34.056 | 45.201 | 57.763 | 71.712 | 87.022 | 12.698 | 19.130 | 26.738 | 35.488 | 45.350 | 56.302 | 68.321 |
| Cs-137 | 661.66 | 16.573 | 24.968 | 34.898 | 46.317 | 59.190 | 73.483 | 89.171 | 13.012 | 19.602 | 27.398 | 36.364 | 46.470 | 57.692 | 70.008 |
| Eu-152 | 778.9 | 16.674 | 25.120 | 35.111 | 46.600 | 59.551 | 73.932 | 89.716 | 13.091 | 19.722 | 27.566 | 36.586 | 46.754 | 58.045 | 70.436 |
| Eu-152 | 964.13 | 16.808 | 25.321 | 35.392 | 46.973 | 60.028 | 74.524 | 90.433 | 13.196 | 19.880 | 27.786 | 36.879 | 47.128 | 58.509 | 71.000 |
| Co-60 | 1173.23 | 16.931 | 25.507 | 35.652 | 47.319 | 60.469 | 75.072 | 91.098 | 13.293 | 20.026 | 27.991 | 37.150 | 47.475 | 58.939 | 71.522 |
| Co-60 | 1332.5 | 17.012 | 25.629 | 35.822 | 47.544 | 60.757 | 75.430 | 91.533 | 13.356 | 20.121 | 28.124 | 37.327 | 47.701 | 59.220 | 71.863 |
| Eu-152 | 1408.01 | 17.047 | 25.682 | 35.896 | 47.642 | 60.883 | 75.585 | 91.721 | 13.384 | 20.163 | 28.182 | 37.404 | 47.799 | 59.342 | 72.011 |
| Nuclide | Energy | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{V} 2}}{\Omega_{\mathrm{P} 10}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 7}}$ | $\frac{\Omega_{\mathrm{v} 2}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{V} 2}}{\Omega_{\mathrm{P} 9}}$ | $\frac{\Omega_{\mathrm{V} 2}}{\Omega_{\mathrm{P} 10}}$ |
| Am-241 | 59.53 | 13.014 | 19.605 | 27.403 | 36.370 | 46.477 | 57.701 | 70.019 | 10.308 | 15.529 | 21.705 | 28.808 | 36.814 | 45.704 | 55.462 |
| Ba-133 | 80.99 | 13.164 | 19.832 | 27.719 | 36.790 | 47.015 | 58.368 | 70.829 | 10.427 | 15.709 | 21.956 | 29.141 | 37.240 | 46.233 | 56.103 |
| Eu-152 | 121.78 | 13.366 | 20.136 | 28.145 | 37.355 | 47.736 | 59.264 | 71.916 | 10.587 | 15.950 | 22.293 | 29.588 | 37.812 | 46.942 | 56.964 |
| Eu-152 | 244.69 | 13.719 | 20.668 | 28.888 | 38.341 | 48.997 | 60.829 | 73.815 | 10.867 | 16.371 | 22.882 | 30.370 | 38.810 | 48.182 | 58.468 |
| Eu-152 | 344.28 | 13.895 | 20.933 | 29.259 | 38.833 | 49.625 | 61.609 | 74.762 | 11.006 | 16.581 | 23.175 | 30.759 | 39.308 | 48.800 | 59.218 |
| Cs-137 | 661.66 | 14.238 | 21.450 | 29.981 | 39.792 | 50.851 | 63.131 | 76.608 | 11.278 | 16.990 | 23.748 | 31.519 | 40.278 | 50.005 | 60.681 |
| Eu-152 | 778.9 | 14.325 | 21.581 | 30.164 | 40.035 | 51.162 | 63.516 | 77.076 | 11.347 | 17.094 | 23.893 | 31.711 | 40.525 | 50.311 | 61.051 |
| Eu-152 | 964.13 | 14.440 | 21.754 | 30.406 | 40.355 | 51.571 | 64.025 | 77.693 | 11.438 | 17.231 | 24.084 | 31.965 | 40.849 | 50.713 | 61.540 |
| Co-60 | 1173.23 | 14.546 | 21.914 | 30.629 | 40.652 | 51.950 | 64.496 | 78.264 | 11.522 | 17.358 | 24.261 | 32.200 | 41.149 | 51.086 | 61.992 |
| Co-60 | 1332.5 | 14.615 | 22.018 | 30.775 | 40.846 | 52.198 | 64.803 | 78.637 | 11.577 | 17.440 | 24.377 | 32.354 | 41.345 | 51.330 | 62.288 |
| Eu-152 | 1408.01 | 14.645 | 22.064 | 30.839 | 40.930 | 52.305 | 64.936 | 78.799 | 11.600 | 17.476 | 24.427 | 32.420 | 41.431 | 51.435 | 62.416 |
| Nuclide | Energy | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 7}}$ | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 10}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 4}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 5}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{P} 6}}$ | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\mathrm{P} 7}}$ | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\mathrm{P} 8}}$ | $\frac{\Omega_{\mathrm{v} 3}}{\Omega_{\mathrm{p} 9}}$ | $\frac{\Omega_{\mathrm{V} 3}}{\Omega_{\mathrm{P} 10}}$ |
| Am-241 | 59.53 | 11.180 | 16.843 | 23.542 | 31.246 | 39.930 | 49.572 | 60.155 | 8.935 | 13.460 | 18.813 | 24.970 | 31.909 | 39.615 | 48.072 |
| Ba-133 | 80.99 | 11.310 | 17.038 | 23.814 | 31.607 | 40.391 | 50.145 | 60.850 | 9.038 | 13.616 | 19.031 | 25.258 | 32.278 | 40.073 | 48.628 |
| Eu-152 | 121.78 | 11.483 | 17.300 | 24.180 | 32.092 | 41.011 | 50.915 | 61.784 | 9.177 | 13.825 | 19.323 | 25.646 | 32.774 | 40.688 | 49.374 |
| Eu-152 | 244.69 | 11.786 | 17.756 | 24.818 | 32.939 | 42.094 | 52.259 | 63.415 | 9.419 | 14.190 | 19.833 | 26.323 | 33.639 | 41.762 | 50.678 |
| Eu-152 | 344.28 | 11.937 | 17.984 | 25.137 | 33.362 | 42.634 | 52.930 | 64.229 | 9.540 | 14.372 | 20.088 | 26.661 | 34.070 | 42.298 | 51.328 |
| Cs-137 | 661.66 | 12.232 | 18.428 | 25.757 | 34.186 | 43.687 | 54.237 | 65.815 | 9.775 | 14.727 | 20.584 | 27.319 | 34.912 | 43.343 | 52.595 |
| Eu-152 | 778.9 | 12.307 | 18.541 | 25.915 | 34.395 | 43.954 | 54.568 | 66.218 | 9.835 | 14.817 | 20.709 | 27.486 | 35.125 | 43.607 | 52.917 |
| Eu-152 | 964.13 | 12.405 | 18.689 | 26.122 | 34.670 | 44.305 | 55.005 | 66.747 | 9.914 | 14.935 | 20.875 | 27.706 | 35.406 | 43.956 | 53.340 |
| Co-60 | 1173.23 | 12.497 | 18.827 | 26.314 | 34.925 | 44.631 | 55.409 | 67.238 | 9.987 | 15.045 | 21.029 | 27.910 | 35.667 | 44.280 | 53.733 |
| Co-60 | 1332.5 | 12.556 | 18.916 | 26.440 | 35.092 | 44.844 | 55.673 | 67.559 | 10.034 | 15.117 | 21.129 | 28.043 | 35.837 | 44.491 | 53.989 |
| Eu-152 | 1408.01 | 12.582 | 18.955 | 26.494 | 35.164 | 44.936 | 55.788 | 67.698 | 10.055 | 15.148 | 21.172 | 28.101 | 35.910 | 44.582 | 54.100 |



Figure 4a: The deviation between the calculated effective solid angle ratio, R , that obtained empirically and the experimental one for D1.


Figure 4b: The deviation between the calculated effective solid angle ratio, R, that obtained empirically and the experimental one for D2.

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